

232°).¹³ Chromium(II) acetate¹⁵ in ethyl alcohol smoothly reduced epoxy ketone 7 (70 mg) to an easily separable mixture of hydroxy ketone 8 (42 mg, prisms from chloroform-methanol, mp 235–239°)¹³ and canarigenone (19 mg). Selective reduction of ketone 8 (30 mg) in ethyl alcohol with Urushibara¹⁶ nickel-A completed synthesis of periplogenin (1c, 26 mg, from methanol, mp 227–234°, lit.⁷ mp 138–232°).

The syntheses of canarigenin, periplogenin, and uzarigenin just described should enhance the availability of these three cardenolides for biological evaluation.

References and Notes

- (1) Part 87: G. R. Pettit and Y. Kamano, *J. Org. Chem.*, in press.
- (2) We are pleased to acknowledge support of this investigation by the National Cancer Institute (performed pursuant to Contract Number NO1-CM-12308, with the Division of Cancer Treatment, National Cancer Institute, Department of Health, Education and Welfare) and Research Grant CA-10612-06 from the National Cancer Institute, the J. W. Kieckhefer Foundation, the Fannie E. Rippel Foundation, The Salt River Project of Arizona, Mrs. Virginia L. Bayless, and The Arizona Public Service Co. We are also grateful to Professor T. Reichstein for the authentic specimen of periplogenin and Dr. W. Haede for the authentic samples of canarigenin and uzarigenin.
- (3) For leading references to cardenolide nomenclature and structural determination by mass spectrometry, refer to P. Brown, F. Bruschweiler, and G. R. Pettit, *Helv. Chim. Acta*, **55**, 531 (1972); P. Brown, F. Bruschweiler, G. R. Pettit, and T. Reichstein, *Org. Mass Spectrometry*, **5**, 573 (1971).
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- (13) Each substance was purified by column chromatography or preparative layer chromatography employing silica gel. Elemental analyses and spectral data (mass, ir, and pmr) were consistent with each structural assignment. The mutual identity of synthetic and authentic samples was established by mixture melting point determination, ir spectral comparison, and thin layer chromatographic comparison.
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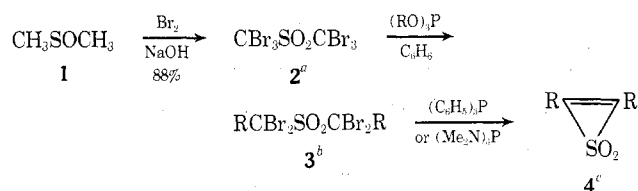
Received April 3, 1974

Synthesis of Alkyl-Substituted Thiirene Dioxides

Summary: A new synthesis of dialkylthiirene 1,1-dioxides *via* debromination of bis(α,α -dibromoalkyl) sulfones by means of trisubstituted phosphines is reported.

Sir: Previously the synthesis of the theoretically interesting and synthetically useful diaryl, aryl alkyl, and dialkylthiirene 1,1-dioxides has been described.¹ Unfortunately, derivatives bearing alkyl substituents have been obtained so far only *via* sulfene and diazoalkane intermediates and therefore are not readily available on a large scale either for an extensive study of their properties or as precursors of other useful synthetic intermediates. With this deficiency in mind we now describe a new route to the dialkyl thiirene dioxides 4 which makes these compounds as easily obtainable as the diaryl analogs. Central to the new approach is the 1,3-elimination² of bromine from a bis(α,α -dibromoalkyl) sulfone by means of triphenylphosphine or a tris(dialkylamino)phosphine [*e.g.*, 3 \rightarrow 4; R = CH₃ (50%); R = CH₃CH₂ (89%)].

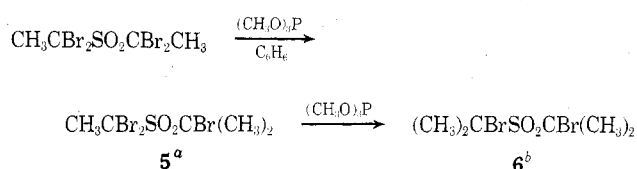
Scheme I



^a Prepared by a modification of the methods of W. V. Farrar [*J. Chem. Soc.*, 508 (1956)] and H. Liebig and H. Pftzting [German Patent 1,256,216; *Chem. Abstr.*, **69**, 18609 (1968)]. ^b R = Et, 70% yield; mp 101.5–102°; nmr (CDCl₃) δ 1.42 (t, 3 H), 3.02 (q, 2 H). ^c R = Et, 89% *via* (Me₂N)₃P at –70°; liquid; nmr (CDCl₃) δ 1.30 (t, 3 H), 2.75 (q, 2 H). Thermolysis (100°) gave 3-hexyne (90%).

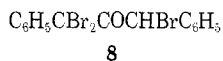
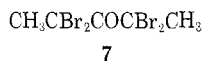
To be useful this method requires a simple synthesis of the tetrabromo sulfones 3, and such a method has now been developed on the basis of a novel but obscure reaction first described in the patent literature.³ Adapting Szabo's method for the synthesis of the corresponding tetrachloro analogs, reaction of trimethyl and triethyl phosphite with bis(tribromomethyl) sulfone (2) gives sulfone 3 (R = Me or Et). In addition to its importance for the preparation of the vinylene sulfones, this unique method of carbon-carbon bond formation combined with conversion to 4 and subsequent facile thermal cycloelimination of sulfur dioxide represents a useful route to internal acetylenes. Although it was not possible to stop the alkylation of 2 by trimethyl phosphite with the introduction of a single methyl group, it was thereafter possible to introduce selectively the third and fourth methyl groups (Scheme II). Tri- and dibromo sulfones such as 5 and 6 are potentially of interest as precursors of olefins by extension of the present process.

Scheme II



^a Yield 80%; mp 126–127°; nmr (CDCl₃) δ 2.48 (s, 6 H), 3.01 (s, 3 H). ^b Yield 90%; mp 129–130°; nmr (CDCl₃) δ 2.30 (s).

This new method can also be applied to the synthesis of cyclopropanones although in the case of the dimethyl derivative the yields obtained so far *via* 7 have been very poor (<2%). On the other hand, treatment of 8⁴ with a 1:1



mixture of tris(dimethylamino)phosphine and triethylamine gave diphenylcyclopropanone in 60% yield. Optimizing the yields in the dialkyl cases would make this a valuable route to the otherwise difficultly accessible aliphatic cyclopropanones.⁶ Sample experimental procedures follow.

Preparation of $\text{CH}_3\text{CBr}_2\text{SO}_2\text{CBr}_2\text{CH}_3$. A solution of 16.6 g of $(\text{MeO})_3\text{P}$ in 30 ml of dry benzene was added dropwise (15 min) to a stirred solution of 37.9 g of $\text{CBr}_3\text{SO}_2\text{CBr}_3$ in 125 ml of benzene. The temperature rose to 80°. After 1 hr at room temperature the solution was washed with water and 10% NaHCO_3 , dried (MgSO_4), and concentrated to 50 ml *in vacuo*. Dilution with 300 ml of cold hexane (0°) and recrystallization of the precipitated solid from benzene-hexane (1:2) gave 20 g (70%) of the sulfone, mp 142–143°, nmr (CDCl_3) δ 3.15 (s, CH_3).

Preparation of Dimethylthiirene Dioxide. A solution of 8.76 g of 3 (R = Me) in 150 ml of dry CH_2Cl_2 was treated (15 min) at –40 to –30° with 10.48 g of triphenylphosphine in 50 ml of CH_2Cl_2 . The solution was warmed to 0° for 1 hr and cooled to –10°; 1.44 g of H_2O was added with stirring. The mixture was allowed to warm to room temperature, the solvent evaporated, and the white solid extracted with six 15-ml portions of H_2O . The combined water extracts were added to 300 ml of CH_2Cl_2 ; the solution was cooled to 0° and solid NaHCO_3 added until gas evolution ceased. Anhydrous MgSO_4 was then added (keeping the temperature below 10°) until the drying agent formed a solid mass. The CH_2Cl_2 was decanted and a fresh portion of CH_2Cl_2 (300 ml) used to extract the residue. Evaporation of the combined CH_2Cl_2 solutions gave 1.2 g (50%) of the vinylene sulfone, mp 100.5–101.5° dec (lit.¹ mp 101–101.5° dec).

Acknowledgment. We wish to thank the National Science Foundation (GP-10152) for their generous support of this work.

References and Notes

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- (2) Compare (a) F. G. Bordwell, B. B. Jarvis, and P. W. R. Corfield, *J. Amer. Chem. Soc.*, **90**, 5298 (1968); (b) B. B. Jarvis, S. D. Dutky, and H. L. Ammon, *ibid.*, **94**, 2136 (1972).
- (3) K. Szabo, U. S. Patent 3,106,585; *Chem. Abstr.*, **60**, 2841b (1964); U. S. Patent 3,294,845; *Chem. Abstr.*, **67**, 11210c (1967). See also C. J. Kelly, Ph.D. Dissertation, Indiana University, Bloomington, Ind., 1969.
- (4) The compound reported in the literature⁵ as $\alpha,\alpha,\alpha',\alpha'$ -tetrabromodibenzyl ketone may in fact be the tribromo derivative 8. In our hands all attempts to prepare the tetrabromo ketone gave only the tribromo derivative.
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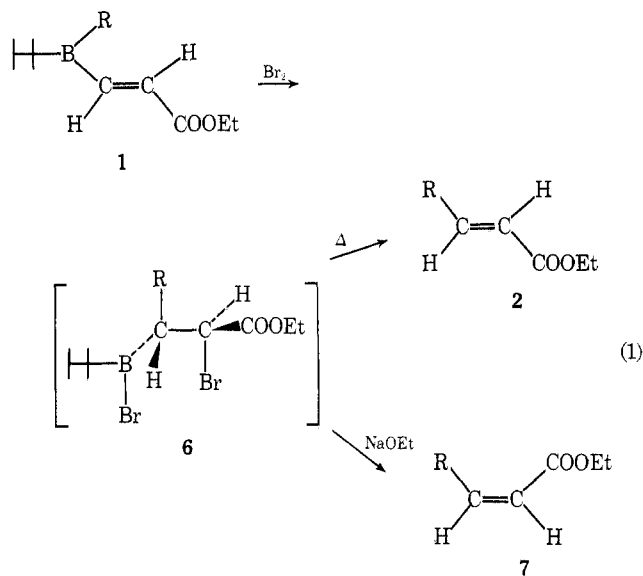
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A Novel Stereoselective Synthesis of (*E*)- and (*Z*)- α,β -Unsaturated Carboxylic Esters *via* Hydroboration

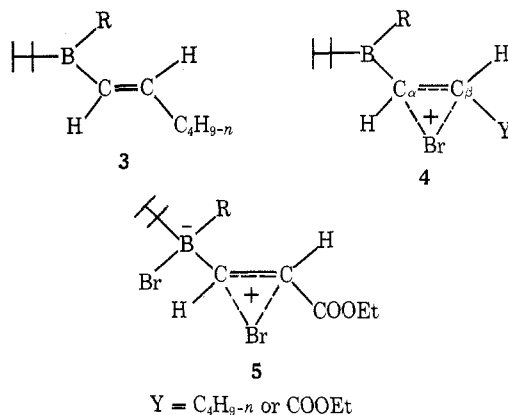
Summary: Treatment of thexylalkyl(β -carboethoxyethenyl)-boranes (1) with 1 equiv of bromine followed by refluxing produces the ethyl esters of (*E*)-3-alkylpropenoic acids (2) in good yields, whereas the bromination of 1 followed by the addition of sodium ethoxide provides the *Z* isomers of 2.

Sir: We wish to report unique results of the bromination of 2,3-dimethyl-2-butylalkylalkenylboranes (thexylalkylalkenylboranes, 1), which permit a convenient stereoselective synthesis of either (*E*)- or (*Z*)- α,β -unsaturated carboxylic esters with the possibility of incorporating stereochemically defined groups in the β position.



We have recently found that the hydroboration of ethyl propiolate, with 2,3-dimethyl-2-butylmonoalkylboranes (thexylmonoalkylboranes)¹ cleanly produces 1.² Treatment of 1 with 1 equiv of bromine followed by refluxing permits coupling of the alkyl group (R) and the β -carboethoxy group to form (*E*)- α,β -unsaturated carboxylic acid esters (2) in a highly stereoselective manner (>95%). The experimental results are summarized in Table I.

The results are unexpected, since the bromination of alkenylboranes are known to produce the corresponding alkenyl bromides.³ Indeed, the bromination of 3 produces predominantly a mixture of (*E*)- and (*Z*)-alkenyl bromides. Therefore, the presence of the β -carboethoxy group must be responsible for diverting the course of the reaction.



Received April 11, 1974